

Copper-Catalyzed Asymmetric Propargylic Substitution Reactions of Propargylic Acetates with Amines**

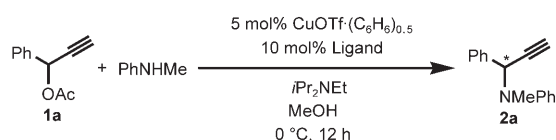
Gaku Hattori, Hiroshi Matsuzawa, Yoshihiro Miyake, and Yoshiaki Nishibayashi*

Recently, we have reported the ruthenium-catalyzed asymmetric propargylic alkylation of propargylic alcohols with acetone to give the corresponding propargylic alkylated products in good yields with a high enantioselectivity (up to 82% *ee*).^[1] This was the first successful example of enantioselective and catalytic propargylic substitution reactions catalyzed by chiral thiolate-bridged diruthenium complexes.^[1] More recently, as an extension of our study, we found the asymmetric propargylation of aromatic compounds with propargylic alcohols catalyzed by the same diruthenium complexes to give the corresponding propargylated aromatic compounds with a high enantioselectivity (up to 94% *ee*).^[2] This method provided a novel approach for the catalytic asymmetric Friedel–Crafts alkylation of aromatic compounds by using propargylic alcohols as a new type of electrophile. However, propargylic substitution reactions with heteroatom-centered nucleophiles, such as alcohols, amines, thiols, and diphenylphosphine oxide, did not proceed enantioselectively in the presence of a catalytic amount of the same chiral diruthenium complexes. These results prompted us to investigate other transition-metal-catalyzed asymmetric propargylic substitution reactions with heteroatom-centered nucleophiles.^[3]

Optically active propargylic amines are synthetically versatile intermediates for the construction of various biologically active compounds and polyfunctional amino derivatives.^[4] Recently, the transition-metal-catalyzed enantioselective addition of terminal alkynes to imines has been developed to produce the corresponding chiral propargylic amines with a high enantioselectivity,^[5] with copper-catalysis being most reliable.^[6] As another synthetic approach to chiral propargylic amines, we have now envisaged developing an

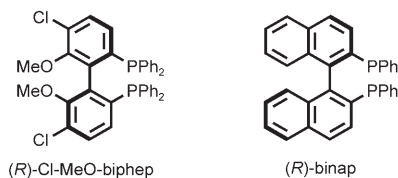
enantioselective version of the copper-catalyzed propargylic substitution reactions of propargylic esters with amines that was originally reported by Murahashi and co-workers.^[7] The first example of the copper-catalyzed asymmetric propargylic substitution reactions with amines has quite recently been reported by using optically active 2,6-bis(oxazolonyl)pyridines (pybox)^[8] as chiral ligands, where the corresponding amines were obtained with up to 88% *ee*.^[9] Herein, we describe a preliminary result of the copper-catalyzed enantioselective propargylic substitution reactions of propargylic acetates with amines by using optically active diphosphines, such as binap^[10] and biphep,^[11] as chiral ligands to afford the corresponding propargylic amines with a high enantioselectivity (up to 89% *ee*).

Treatment of 1-phenyl-2-propynyl acetate (**1a**) with *N*-methylaniline and *N,N*-diisopropylethylamine^[12] in the presence of a catalytic amount of copper trifluoromethanesulfonate benzene complex, CuOTf·(benzene)_{0.5} (5 mol%) and (*R*)-Cl-MeO-biphep^[13,14] (10 mol%) in methanol^[15] at 0 °C for 12 h gave *N*-methyl-*N*-(1-phenyl-2-propynyl)aniline (**2a**) quantitatively with 85% *ee* (Scheme 1). The use of an equal amount of (*R*)-Cl-MeO-biphep (5 mol%) to the copper complex slightly decreased the enantioselectivity of **2a**. When (*R*)-binap was used in place of (*R*)-Cl-MeO-biphep as the chiral diphosphine, the reaction proceeded sluggishly with a similar enantioselectivity.^[16]



Ligand	Yield of 2a	<i>ee</i> of 2a
(<i>R</i>)-Cl-MeO-biphep	96%	85% <i>ee</i>
(<i>R</i>)-Cl-MeO-biphep ^[a]	93%	81% <i>ee</i>
(<i>R</i>)-binap	82%	85% <i>ee</i>

[a] 5 mol% of (*R*)-Cl-MeO-biphep was used.



Scheme 1. Reactions of 1-phenyl-2-propynyl acetate (**1a**) with *N*-methylaniline in the presence of a catalytic amount of copper trifluoromethanesulfonate benzene complex and (*R*)-Cl-MeO-biphep or (*R*)-binap.

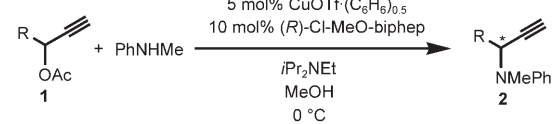
[*] G. Hattori, Dr. H. Matsuzawa, Dr. Y. Miyake, Prof. Dr. Y. Nishibayashi
Institute of Engineering Innovation
School of Engineering
The University of Tokyo
Yayoi, Bunkyo-ku, Tokyo, 113-8656 (Japan)
Fax: (+81) 3-5841-1175
E-mail: ynishiba@sogo.t.u-tokyo.ac.jp
Homepage: <http://park.itc.u-tokyo.ac.jp/nishiba/>

[**] This work was supported by Grant-in-Aids for Scientific Research for Young Scientists (S) (No. 19675002) and for Scientific Research on Priority Areas (No. 18066003) from the Ministry of Education, Culture, Sports, Science and Technology (Japan). G.H. acknowledges the Global COE Program for Chemistry Innovation. Y.N. also thanks to the Iketani Science & Technology Foundation and the Japan Securities Scholarship Foundation. We thank Mr. Yusuke Shimada for some experiments at the initial stage of this work.

Supporting information for this article is available on the WWW under <http://www.angewandte.org> or from the author.

Next, the catalytic propargylic amination of other propargylic acetates (**1**) was investigated by using (*R*)-Cl-MeO-biphep as a chiral ligand. Typical results are shown in Table 1.

Table 1: Copper-catalyzed enantioselective propargylic amination of propargylic acetates (**1**) with *N*-methylaniline.^[a]



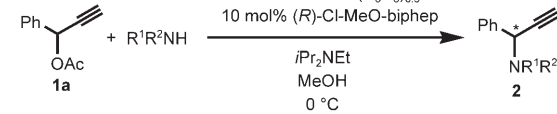
Entry	R	t [h]	Yield of 2 [%] ^[b]	ee of 2 [%] ^[c]
1	1a , Ph	12	2a , 96	85
2	1b , <i>p</i> -MeC ₆ H ₄	12	2b , 98	81
3	1c , <i>p</i> -MeOC ₆ H ₄	12	2c , 99	82
4	1d , <i>p</i> -ClC ₆ H ₄	24	2d , 94	85
5	1e , <i>p</i> -BrC ₆ H ₄	36	2e , 91	85
6	1f , <i>p</i> -PhC ₆ H ₄	12	2f , 99	84
7	1g , 1-naphthyl	12	2g , 99	85
8	1h , 2-naphthyl	12	2h , 99	83
9	1i , cyclohexyl	48	2i , 0	–

[a] All the reactions of **1** (0.20 mmol) with *N*-methylaniline (0.40 mmol) and *N,N*-diisopropylethylamine (0.80 mmol) were carried out in the presence of copper trifluoromethanesulfonate benzene complex (0.010 mmol) and (*R*)-Cl-MeO-biphep (0.020 mmol) in MeOH (2.0 mL) at 0 °C. [b] Yield of isolated product. [c] Determined by HPLC (see Supporting Information for experimental details).

The introduction of substituents such as methyl, methoxy, chloro, bromo, and phenyl, at the *para*-position in the benzene ring of the propargylic acetate **1a** did not appreciably effect the yield and enantioselectivity of the products (**2**; Table 1, entries 1–6). Enantioselectivity of the same level was also observed in the reactions of 1-naphthyl-2-propynyl acetates (**1g** and **1h**) with *N*-methylaniline under the same reaction conditions (Table 1, entries 7 and 8). After recrystallization, optically pure propargylic amines were obtained in some cases. The presence of an aryl group at the propargylic position of the propargylic acetate was necessary to achieve a high enantioselectivity, and the reaction of 1-cyclohexyl-2-propynyl acetate (**1i**) with *N*-methylaniline did not proceed at all under the same reaction conditions (Table 1, entry 9). Further, the also reaction did not proceed when propargylic acetate bearing an internal alkyne moiety, such as 1,3-diphenyl-2-propynyl acetate, was used as a substrate, indicating that only the propargylic acetates bearing a terminal alkyne moiety can be employed successfully for this reaction.

A variety of amines are available as nitrogen-centered nucleophiles. Typical results are shown in Table 2. The introduction of substituents in the benzene ring of *N*-methylaniline slightly affected the enantioselectivity of the product. In fact, the presence of a chloro moiety at *para*-position of the benzene ring slightly increased the enantioselectivity, but that of a methyl group at *para*-position slightly decreased it (Table 2, entries 1–4). The use of cyclic amines such as piperidine and 1,2,3,4-tetrahydroquinoline as nucleophiles slightly decreased the enantioselectivity (Table 2, entries 5 and 6). A similar enantioselectivity was achieved when *N*-ethylaniline was used (Table 2, entry 7). In sharp contrast to the copper-pybox system^[9] with which the

Table 2: Copper-catalyzed enantioselective propargylic amination of 1-phenyl-2-propynyl acetate (**1a**) with amines.^[a]

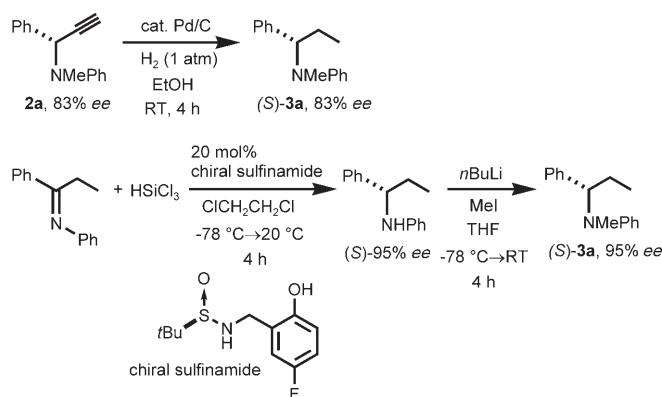


Entry	R ¹ , R ²	t [h]	Yield of 2 [%] ^[b]	ee of 2 [%] ^[c]
1	Ph, Me	12	2a , 96	85
2	<i>p</i> -ClC ₆ H ₄ , Me	12	2j , 96	89
3 ^[d]	<i>p</i> -ClC ₆ H ₄ , Me	12	2k , 92	89
4	<i>p</i> -MeC ₆ H ₄ , Me	12	2l , 97	82
5	-(CH ₂) ₅ -	12	2m , 64	80
6	-C ₆ H ₄ (CH ₂) ₃ - ^[e]	12	2n , 97	81
7	Ph, Et	12	2o , 95	83
8	Ph, H	12	2p , 90	53

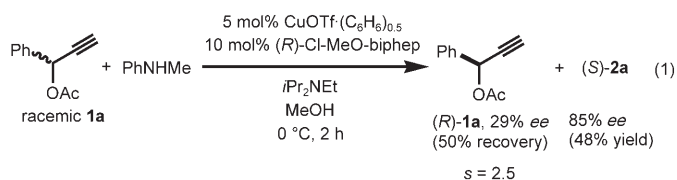
[a] All the reactions of **1a** (0.20 mmol) with amines (0.40 mmol) and *N,N*-diisopropylethylamine (0.80 mmol) were carried out in the presence of copper trifluoromethanesulfonate benzene complex (0.010 mmol) and (*R*)-Cl-MeO-biphep (0.020 mmol) in MeOH (2.0 mL) at 0 °C. [b] Yield of isolated product. [c] Determined by HPLC (see Supporting Information for experimental details). [d] **1d** was used in place of **1a**. [e] 1,2,3,4-Tetrahydroquinoline was used as an amine.

reactions with aniline afforded the highest enantioselectivity of the products, only a low enantioselectivity was observed in our case (Table 2, entry 8).

To obtain more information on the enantioselective propargylic amination, the stereochemistry of the product **2a** was determined. Hydrogenation of **2a** in the presence of a catalytic amount of Pd/C under 1 atm of H₂ at room temperature for 4 h gave *N*-methyl-*N*-(1-phenylpropyl)aniline (**3a**; Scheme 2). The absolute configuration was in accord with (*S*)-**3a** obtained by the reported method,^[17,18] showing that the original propargylic amine **2a** has an *S* absolute configuration. In a separate experiment, a low enantioselectivity was observed in **1a** after recovered a short reaction time as shown in Equation (1). This result means that there is a small difference in the reactivity between (*R*)-**1a** and (*S*)-**1a**, the latter being slightly more reactive. Although the detailed reaction pathway is not yet clear, we consider that *N*-methylaniline might attack the copper acetylide complex, which has a cationic γ -carbon atom, from the *re* face. The



Scheme 2. Determination of the absolute configuration of propargylic amine (**2a**).



copper acetylide complex is formed in situ from **1a** and the copper complex bearing (*R*)-Cl-MeO-biphep.

In summary, we have found the copper-catalyzed enantioselective propargylic substitution reactions of propargylic acetates with amines to give the corresponding propargylic amines in excellent yields with a high enantioselectivity (up to 89% *ee*). Optically active diphosphines such as Cl-MeO-biphep and binap have been found to function as suitable chiral ligands. The reaction described may provide a novel synthetic method for the preparation of chiral propargylic amines. Further investigation of the reaction mechanism^[19] and for broadening the synthetic application of this asymmetric propargylic amination is in progress.

Experimental Section

A typical experimental procedure for the reaction of 1-phenyl-2-propynyl acetate (**1a**) with *N*-methylaniline: CuOTf·(benzene)_{0.5} (2.5 mg, 0.010 mmol) and (*R*)-Cl-MeO-biphep (13.0 mg, 0.020 mmol) were placed in a 20 mL round-bottomed flask under N₂. Anhydrous methanol (1.0 mL) was added, and then the mixture was stirred at 60 °C for 1 h with a magnetic stirrer. Then, **1a** (34.8 mg, 0.20 mmol), *N*-methylaniline (42.8 mg, 0.40 mmol), and diisopropylethylamine (103.3 mg, 0.80 mmol) in anhydrous methanol (1.0 mL) were added under N₂, and the reaction flask was kept at 0 °C for 12 h. The solvent was concentrated under reduced pressure by an aspirator, and the residue was purified by the column chromatography (SiO₂) with hexane and ethyl acetate (97:3) as eluent to give *N*-methyl-*N*-(1-phenyl-2-propynyl)aniline (**2a**)^[20] as a pale yellow solid (42.5 mg, 0.19 mmol; 96% isolated yield). [α]_D²² = +10.7 (*c* 1.42, CHCl₃). The optical purity of **2a** was determined by HPLC analysis; DAICEL Chiralpak AD, hexane/*i*PrOH = 97/3, flow rate = 1.0 mL min⁻¹, λ = 254 nm, retention time; 5.64 min (minor) and 7.04 min (major), 85% *ee*.

Received: January 18, 2008

Published online: April 11, 2008

Keywords: amination · asymmetric catalysis · copper · nucleophilic substitution · P ligands

- [1] Y. Inada, Y. Nishibayashi, S. Uemura, *Angew. Chem.* **2005**, *117*, 7893; *Angew. Chem. Int. Ed.* **2005**, *44*, 7715.
 [2] a) H. Matsuzawa, Y. Miyake, Y. Nishibayashi, *Angew. Chem.* **2007**, *119*, 6608; *Angew. Chem. Int. Ed.* **2007**, *46*, 6488; b) H. Mastuzawa, K. Kanao, Y. Miyake, Y. Nishibayashi, *Org. Lett.* **2007**, *9*, 5561.
 [3] A variety of transition-metal-catalyzed propargylic substitution reactions of propargylic alcohol derivatives with heteroatom-centered nucleophiles have been reported: a) J. A. Marshall, M. A. Wolf, *J. Org. Chem.* **1996**, *61*, 3238; b) R. Mahrwald, S. Quint, *Tetrahedron* **2000**, *56*, 7463; c) R. Mahrwald, S. Quint, *Tetrahedron Lett.* **2001**, *42*, 1655; d) B. D. Sherry, A. T. Radosevich, F. D. Toste, *J. Am. Chem. Soc.* **2003**, *125*, 6076; e) M. Georgy, V. Boucard, J.-M. Campagne, *J. Am. Chem. Soc.* **2005**,

- 127*, 14180; f) P. A. Evans, M. J. Lawler, *Angew. Chem.* **2006**, *118*, 5092; *Angew. Chem. Int. Ed.* **2006**, *45*, 4970; g) Z.-P. Zhan, W.-Z. Yang, R.-F. Yang, J.-L. Yu, J.-P. Li, H.-J. Liu, *Chem. Commun.* **2006**, 3352; h) Z.-P. Zhan, J.-L. Yu, H.-J. Liu, Y.-Y. Cui, R.-F. Yang, W.-Z. Yang, J.-P. Li, *J. Org. Chem.* **2006**, *71*, 8298; i) H. Qin, N. Yamagiwa, S. Matsunaga, M. Shibasaki, *Angew. Chem.* **2007**, *119*, 413; *Angew. Chem. Int. Ed.* **2007**, *46*, 409.
 [4] For examples, see: a) J. A. Porco, Jr., F. J. Schoene, T. J. Stout, J. Clardy, S. L. Schreiber, *J. Am. Chem. Soc.* **1990**, *112*, 7410; b) K. C. Nicolaou, C.-K. Hwang, A. L. Smith, S. V. Wendeborn, *J. Am. Chem. Soc.* **1990**, *112*, 7416; c) P. H. Yu, B. A. Davis, A. A. Boulton, *J. Med. Chem.* **1992**, *35*, 3705; d) B. Jiang, M. Xu, *Angew. Chem.* **2004**, *116*, 2597; *Angew. Chem. Int. Ed.* **2004**, *43*, 2543; e) Y. Yamamoto, H. Hayashi, T. Saigoku, H. Nishiyama, *J. Am. Chem. Soc.* **2005**, *127*, 10804; f) J. J. Fleming, J. D. Bois, *J. Am. Chem. Soc.* **2006**, *128*, 3926.
 [5] For a recent review, see: L. Zani, C. Bolm, *Chem. Commun.* **2006**, 4263, and references therein.
 [6] For examples, see: a) C. Wei, C.-J. Li, *J. Am. Chem. Soc.* **2002**, *124*, 5638; b) C. Koradin, K. Polborn, P. Knochel, *Angew. Chem.* **2002**, *114*, 2651; *Angew. Chem. Int. Ed.* **2002**, *41*, 2535; c) N. Gommermann, C. Koradin, K. Polborn, P. Knochel, *Angew. Chem.* **2003**, *115*, 5941; *Angew. Chem. Int. Ed.* **2003**, *42*, 5763; d) T. F. Knöpfel, P. Aschwanden, T. Ichikawa, T. Watanabe, E. M. Carreira, *Angew. Chem.* **2004**, *116*, 6097; *Angew. Chem. Int. Ed.* **2004**, *43*, 5971.
 [7] a) Y. Imada, M. Yuasa, I. Nakamura, S.-I. Murahashi, *J. Org. Chem.* **1994**, *59*, 2282; b) R. Geri, C. Oilizzi, L. Lardicci, A. M. Caporusso, *Gazz. Chim. Ital.* **1994**, *124*, 241.
 [8] H. Nishiyama, Y. Itoh, H. Matsumoto, S.-B. Park, K. Itoh, *J. Am. Chem. Soc.* **1994**, *116*, 2223.
 [9] a) R. J. Detz, M. M. E. Delville, H. Hiemstra, J. H. van Maarseveen, *Angew. Chem.* **2008**, *120*, 3837; *Angew. Chem. Int. Ed.* **2008**, *47*, 3777; b) Prof. Maarseveen and his co-workers presented a part of the result at the PAC-Symposium 2007 (June 15, 2007, Amsterdam).
 [10] Binap = 2,2'-bis(diphenylphosphino)-1,1'-binaphthyl.
 [11] Biphep = 2,2'-bis(diphenylphosphino)-1,1'-biphenyl.
 [12] No reaction occurred in the absence of *N,N*-diisopropylethylamine. When triethylamine was used in place of *N,N*-diisopropylethylamine as a base, a similar enantioselectivity was observed.
 [13] (*R*)-Cl-MeO-biphep = (*R*)-5,5'-dichloro-6,6'-dimethoxy-2,2'-bis(diphenylphosphino)-1,1'-biphenyl, is a commercially available reagent from Strem Chemicals.
 [14] a) R. R. Huddleston, H.-Y. Jang, M. J. Krische, *J. Am. Chem. Soc.* **2003**, *125*, 11488; b) H.-Y. Jang, F. W. Hughes, H. Gong, J. Zhang, J. S. Brodbelt, M. J. Krische, *J. Am. Chem. Soc.* **2005**, *127*, 6174; c) J. U. Rhee, M. J. Krische, *J. Am. Chem. Soc.* **2006**, *128*, 10674.
 [15] The reaction did not proceed smoothly in other solvents such as EtOH, CH₂Cl₂, toluene, THF, MeCN, DMF, and 1,4-dioxane.
 [16] Although a variety of optically active diphosphines and monophosphines, such as Duphos, chiraphos, diop, and mop, were investigated as chiral ligands for the propargylic amination, these did not work effectively. As a result, (*R*)-Cl-MeO-biphep was found to be the ligand of choice for the present propargylic amination. The detailed experimental results will be reported in due course.
 [17] D. Pei, Z. Wang, S. Wei, Y. Zhang, J. Sun, *Org. Lett.* **2006**, *8*, 5913.
 [18] A. M. Johns, N. Sakai, A. Ridder, J. F. Hartwig, *J. Am. Chem. Soc.* **2006**, *128*, 9306.
 [19] The molecular structure of a copper-(*R*)-Cl-MeO-biphep complex has been determined by X-ray crystallography. The detailed experimental results will be reported in due course.
 [20] Y. Nishibayashi, M. D. Milton, Y. Inada, M. Yoshikawa, I. Wakiji, M. Hidai, S. Uemura, *Chem. Eur. J.* **2005**, *11*, 1433.